1. and 2.

$$F = G \underset{V^2}{M_1} \underset{V^2}{M_2} \xrightarrow{V \to 6} \underset{V^2}{M_1} \xrightarrow{V_2} \underset{W_1}{W_2} \xrightarrow{V_2} \underset{W_2}{W_2} \xrightarrow{V \to 6} \xrightarrow{W_1} \underbrace{V_2}_{W_2} \xrightarrow{V_2} \underbrace{V \to 6} \xrightarrow{W_1} \underbrace{V \to 6} \xrightarrow{W_1} \underbrace{V \to 6} \xrightarrow{W_2} \underbrace{V \to 6} \xrightarrow{W_1} \underbrace{V \to 6} \xrightarrow{V \to 6} \underbrace{V \to 6} \underbrace{V \to 6} \xrightarrow{V \to 6} \underbrace{V \to 6} \underbrace{V$$

[5 marks] and [5 marks]

3. Given the adiabatic law $p = K \rho^{\gamma}$,

where *K* is a constant, then

$$u^{2} = \left[\frac{\partial p}{\partial \rho}\right]_{\text{adiabatic}} = \frac{\partial}{\partial \rho} \left(K\rho^{\gamma}\right) = \gamma K\rho^{\gamma-1}$$

Hence

 $u = (\gamma K)^{1/2} \rho^{(\gamma - 1)/2}.$ [3 marks] 4. We have $\rho = 2n_{\text{H}_2}m_{\text{H}},$ where $n_{\rm H_2}$ is the number-density of hydrogen molecules and $m_{\rm H}$ is the mass of the hydrogen atom. Hence

$$\tau_{\rm ff} \sim (2Gn_{\rm H_2}m_{\rm H})^{-1/2}$$

= $[2 \times 6.67 \times 10^{-11} \times 10^{10} \times 1.67 \times 10^{-27}]^{-1/2}$
 $\approx 2 \times 10^{13} \text{ s} \approx 10^{6} \text{ y.}$
[3 marks]

5. If

$$k < \left(\frac{4\pi G\rho}{u^2}\right)^{1/2},$$

then we have from the dispersion relation $\omega^2 < 0$,

so that ω is imaginary. Hence writing

 $\omega = \pm i \alpha$,

where α is a real positive number and substituting into the wave-like solution for the density perturbation $\rho_1(t,x)$

$$\rho_1(t, x) = \rho_{10} e^{i(\omega t - kx)}$$
$$= \rho_{10} e^{\mp \alpha t} e^{-ikx}$$

The second line of the latter equation shows that we have a standing wave with amplitude

 $\rho_{10}e^{\mp\alpha t}$. [5 marks]

[Total marks available 21]